Atomic Emission Spectroscopy (Chapter 9/10):

Identification of elements but not compounds

Excitation and Atomization:

Traditionally based on

• flame

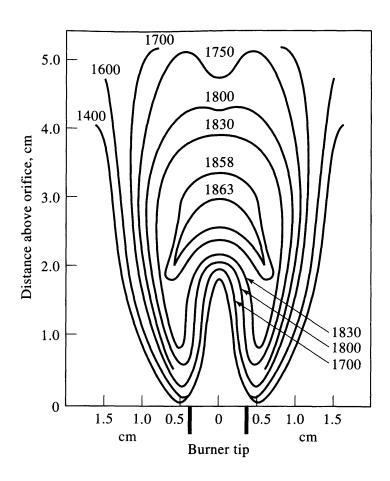
but

- arc and spark
- plasma

excitation offers

- (i) increased atomization/excitation
- (ii) wider range of elements
- (iii) emission from multiple species simultaneously
- (iv) wide dynamic range

Flame Excitation Sources:



(Fig 9-3)

Primary Combustion Zone

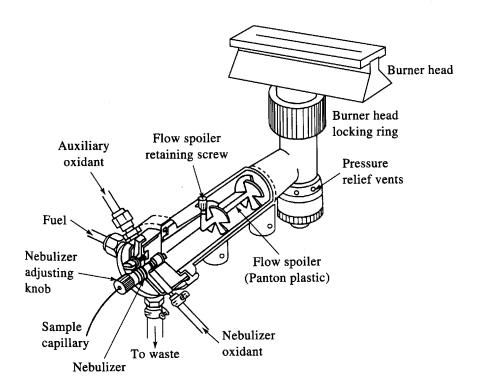
Interzonal Region

Secondary Combustion Zone

Flame Temperatures:

Fuel	Oxidant	Temperature
Gas	Air	~1800 °C
H_2	O_2	~2600 °C
Acetylene	O_2	~3000 °C

Laminar Flow Burner: (Fig 9-5)

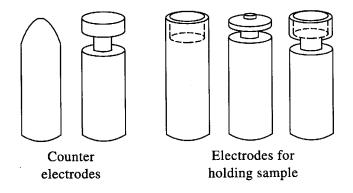


- Cheap
- Simple
- Flame stability
- Low temperature

Arc and Spark Excitation Sources:

- Limited to semiquantitative/qualitative analysis (arc flicker)
- Usually performed on solids
- Largely displaced by plasma-AES

Electric current flowing between two C electrodes (Fig 10-16)



Sample pressed into electrode or mixed with Cu powder and pressed - briquetting

Cyanogen bands (CN) 350-420 nm occur with C electrodes in air - He, Ar atmosphere

Arc/spark unstable - each line measured >20 s (needs multichannel detection)

photographic film:

- Cheap
- Long integration times
- Difficult to develop/analyze
- Non-linearity of line "darkness"

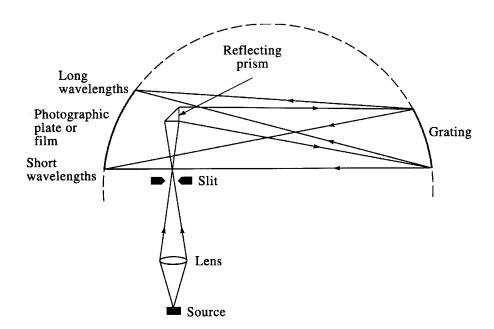


Fig 10-17

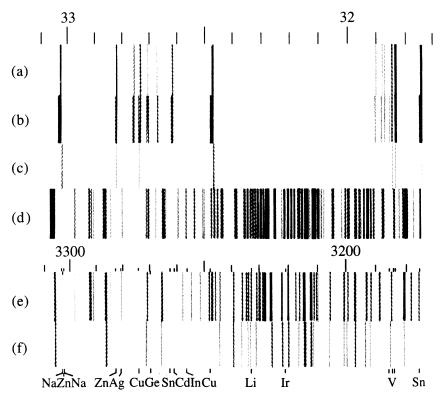


Fig 10-18

multichannel PMT instruments:

- for rapid determinations (<20 lines) but not versatile
- routine analysis of solids metals, alloys, ores, rocks, soils
- portable instruments

Plasma Excitation Sources:

gas containing high proportion of cations and electrons

(1) Inductively Coupled Plasma (ICP)

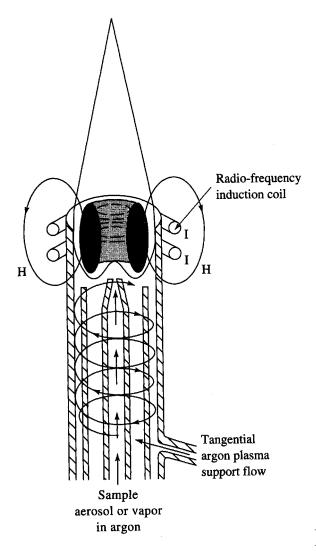
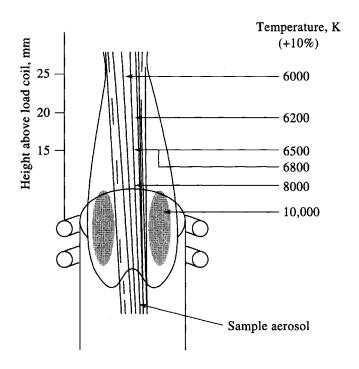


Fig 10-1

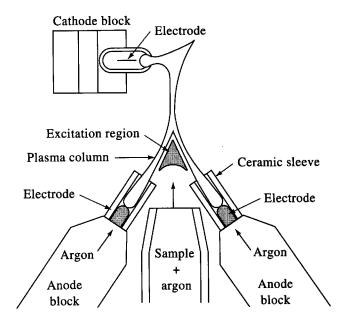
- Torch up to 1" diameter
- Ar cools outer tube, defines plasma shape
- Radio-frequency (RF) up to 2 kW
- Ar flow up to 20 L/min

Plasma Structure: (Fig. 10-4)



- Brilliant white core Ar continuum and lines
- Flame-like tail up to 2 cm
- Transparent region measurements made
- Hotter than flame (10,000 K) more complete atomization/excitation
- Atomized in "inert" atmosphere
- Little ionization too many electrons in plasma

(2) Direct Current (DC) Plasma (Fig 10-5)



- DC current (10-15 A) flows between C anodes and W cathode
- Plasma core at 10,000 K, viewing region at ~5,000 K
- Simpler, less Ar than ICP less expensive

Atomic Emission Spectrometers:

May be >1,000 visible lines (<1 Å) on continuum

Need

- high resolution (<0.1 Å)
- high throughput
- low stray light
- wide dynamic range (>106)
- precise and accurate wavelength calibration/intensities
- stability
- computer controlled

Three instrument types:

sequential (scanning and slew-scanning)

multichannel

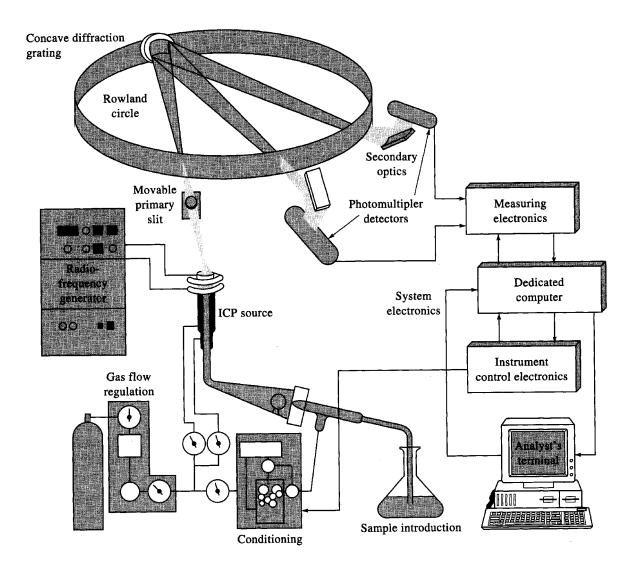
(Fourier transform FT-AES)

Sequential monochromators:

Slew-scan spectrometers - even with many lines, much spectrum contains no information

- rapidly scanned (slewed) across blank regions
- slowly scanned across lines
- computer control/preselected lines to scan

Multichannel AES: (Fig. 10-8)



Sequential instrument - PMT moved behind aperture plate, or grating+prism moved to focus new λ on exit slit

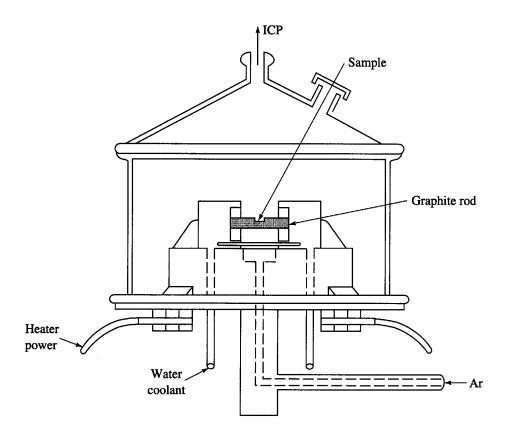
- Cheaper
- Slower
- Pre-configured exit slits to detect up to 20 lines, slew scan

Multichannel instrument - multiple PMT's

- Expensive
- Faster

Solution Sample Introduction:

- (1) Electrothermal vaporizer* (ETV)
 - electric current rapidly heats crucible containing sample
 - sample carried to atomizer by gas (Ar, He)
 - only for introduction, not atomization (Fig 10-3)



- (2) Nebulizer convert solution to fine spray or aerosol
 - (a) Ultrasonic nebulizer uses ultrasound waves to "boil" solution flowing across disc
 - (b) Pneumatic nebulizer uses high pressure gas to entrain solution

Cross-flow Nebulizer

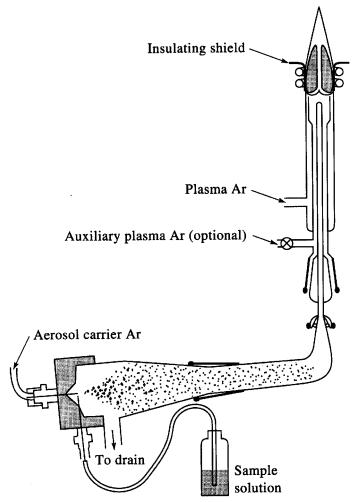


Fig 10-2

Solid Sample Introduction:

- (1) Electrothermal vaporizer*
- (2) Direct Insertion(*) uses powder placed inside flame, plasma, arc or spark atomizer (atomizer acts as vaporizer)
 - Coating on electrode in atomizer
- (3) Ablation uses coating of electrodes in discharge cell and sample entrained in Ar or He gas
 - Laser ablation uses laser to vaporize sample

^{*} intermittent

Applications of AES:

AES relatively insensitive (small excited state population at moderate temperature)

AAS still used more than AES

- (i) less expensive/complex instrumentation
- (ii) lower operating costs
- (iii) greater precision

In practice ~60 elements detectable

- 10 ppb range most metals
- Li, K, Rb, Cs strongest lines in IR
- Large # of lines, increase chance of overlap

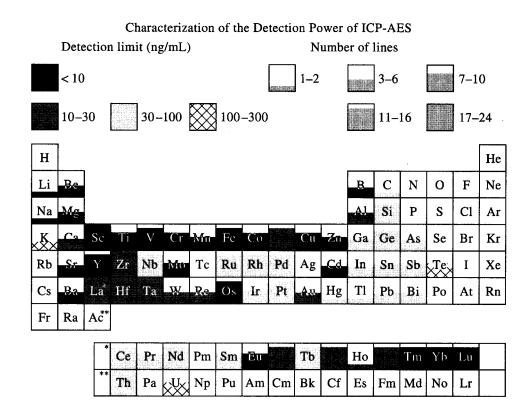


Fig 10-13