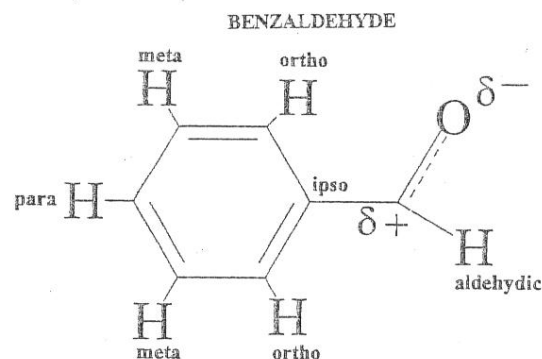
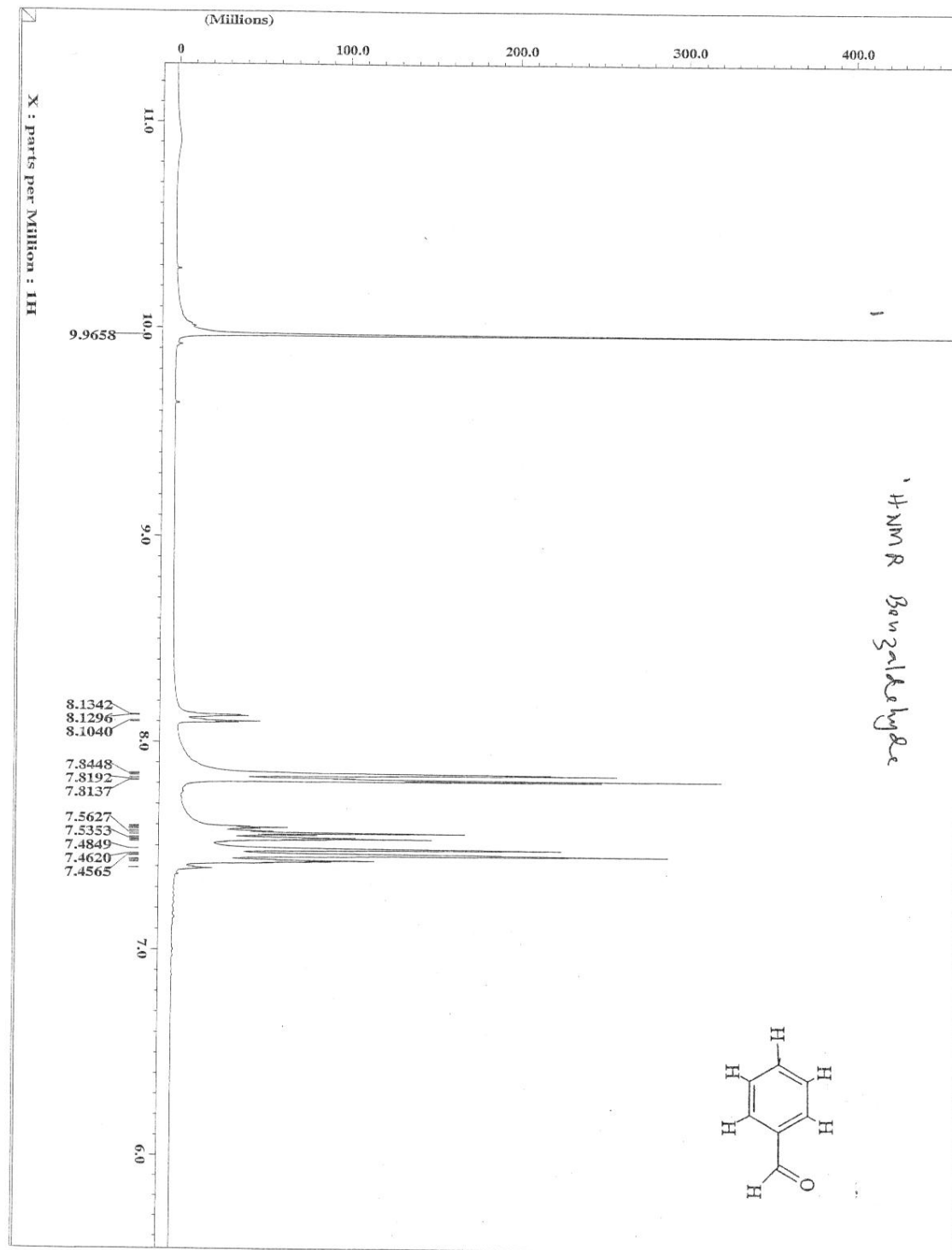


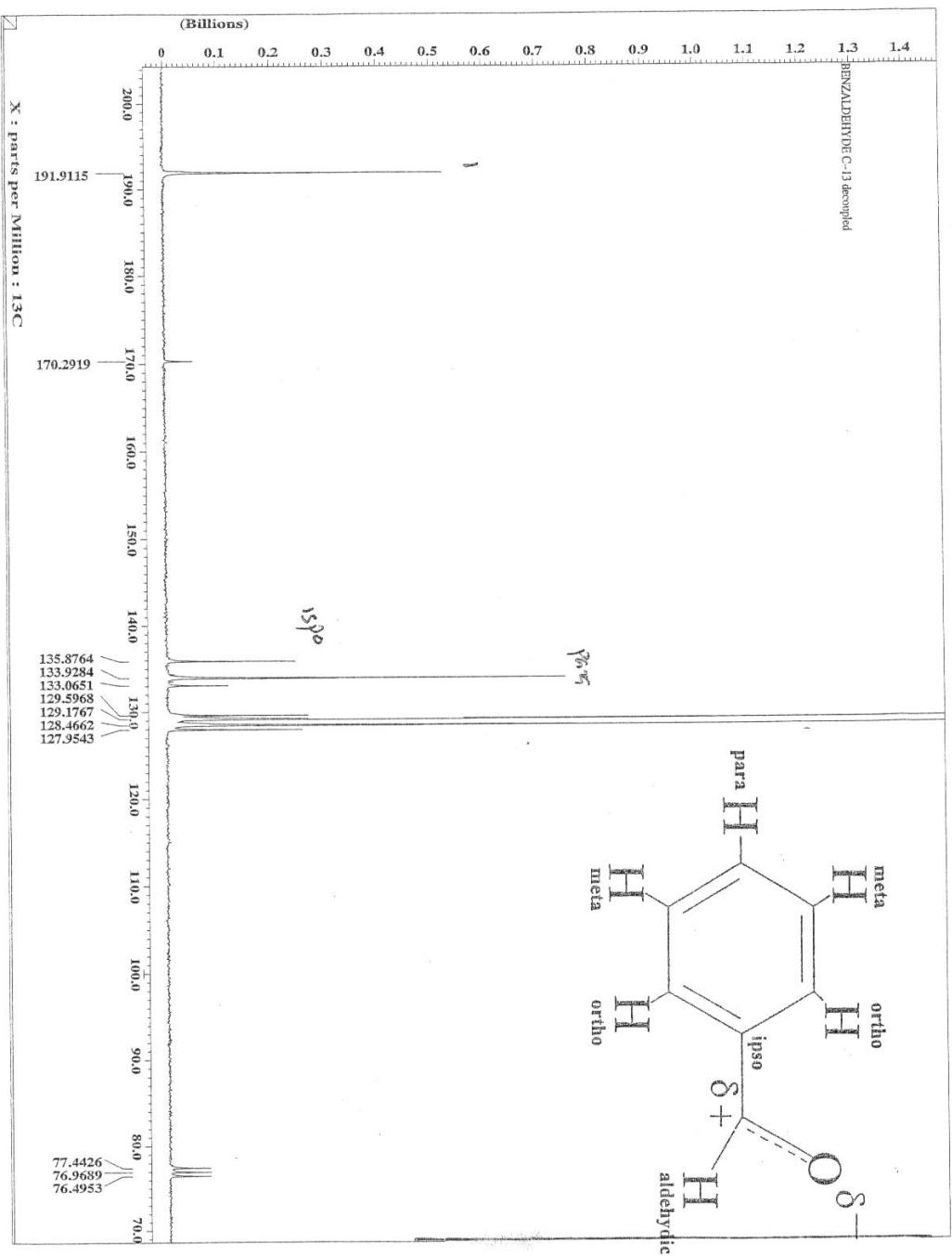
NMR SPECTRAL DATA

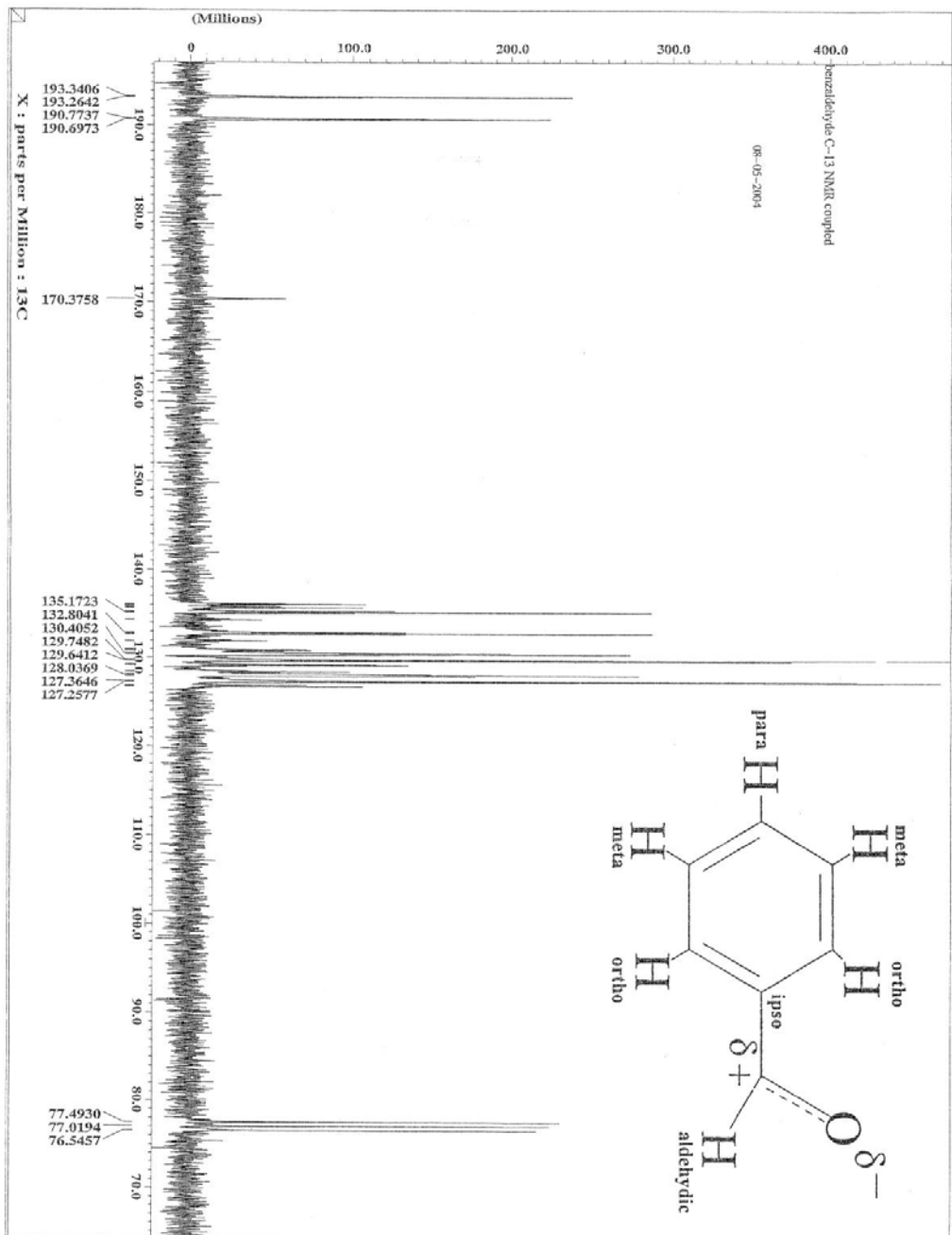
- A. Correlation Data for ^1H and ^{13}C NMR Spectra
- For different kinds of protons.
 - For different kinds of carbons,
- B. How many kinds of chemically nonequivalent protons are present in benzaldehyde?
- This question is answered by visual inspection of structure (

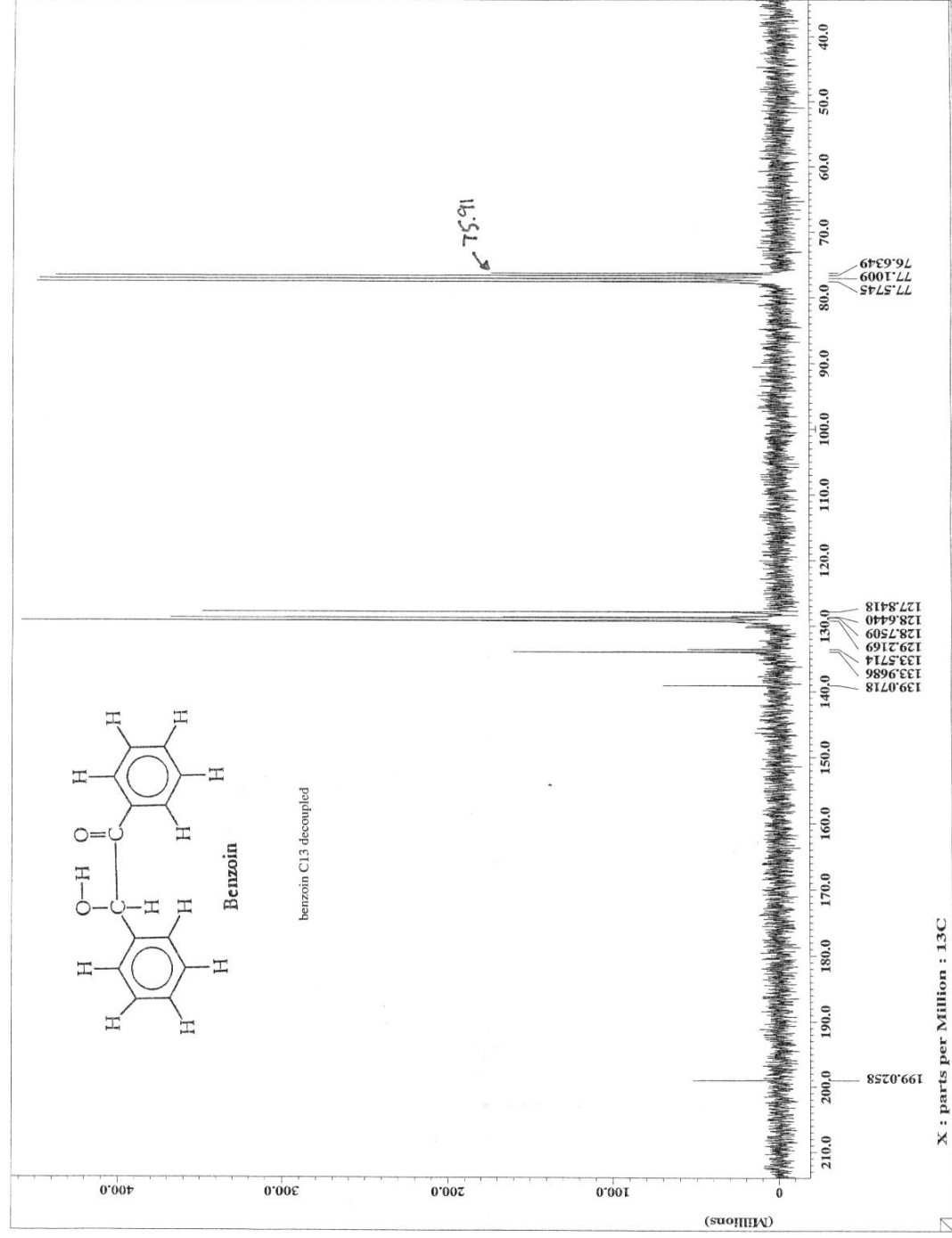


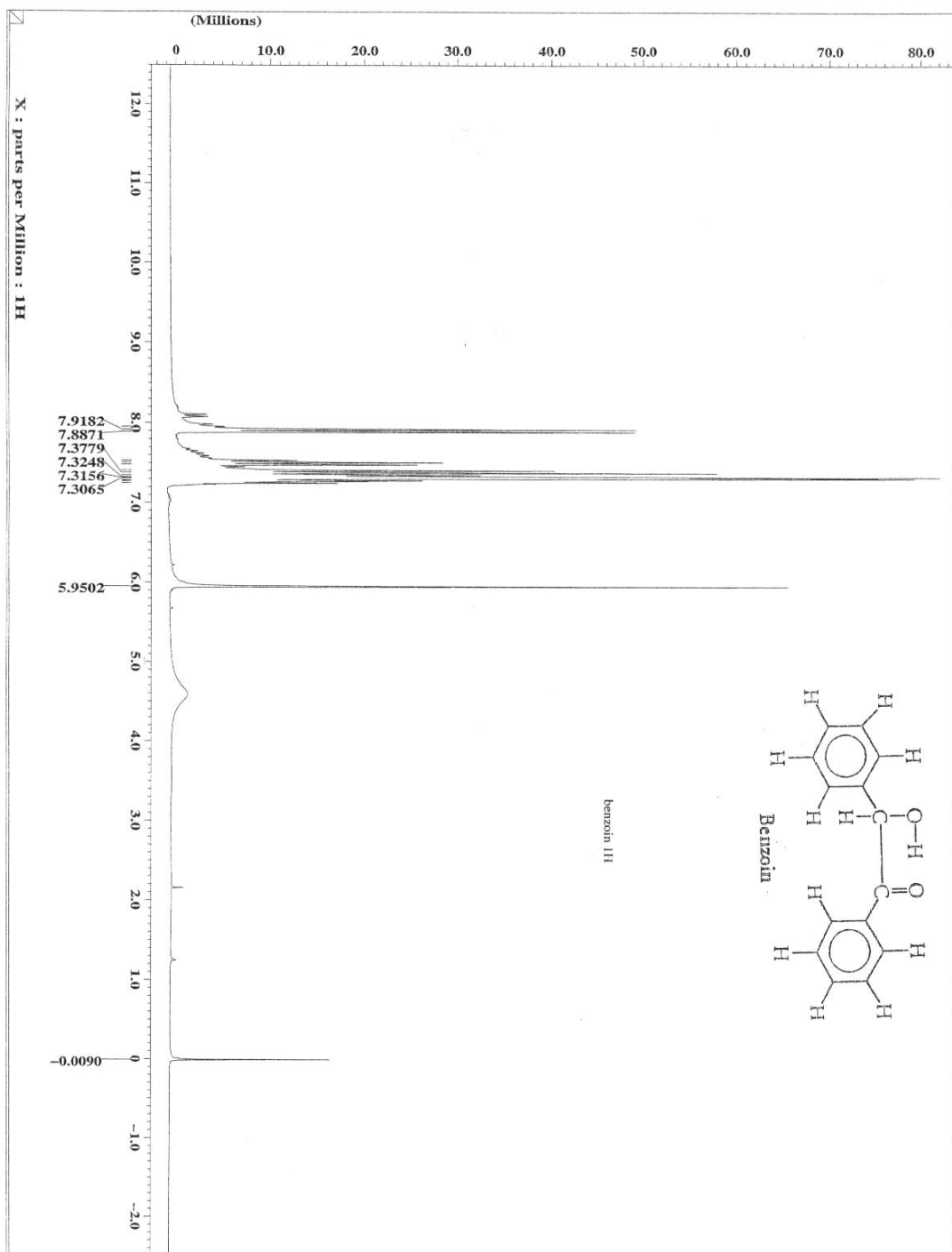
- Carrying out such a visual inspection of the above structure shows four kinds of chemically nonequivalent proton. They are: 2 *meta*-protons, 1 *para*-proton, 2 *ortho*-protons, and 1 aldehyde (or formyl) proton.
- C. Prediction of the chemical shift position and multiplicity for each of the different kinds of chemically nonequivalent protons present in benzaldehyde.
- Table A shows that the aromatic protons come into resonance in the range 6.0-9.5 δ (NOTE: deshielding by the electron-withdrawing formyl group is greatest at the *ortho*-positions, then the *para*-position, and least at the *meta*-positions), and since the signals for the coupling aromatic-protons are close together, a complex absorption peak will arise for each kind of nonequivalent aromatic-proton; Table A also shows that the aldehyde proton comes into resonance in the range 9.5-10.0 δ , and since it is not coupled with any adjacent nonequivalent protons, it will appear as a single peak.
- D. How many kinds of chemically nonequivalent carbon atoms are present in benzaldehyde? Predict the chemical shift position for each one of the different kinds of carbon atoms.
- Visual inspection of the above structure reveals five kinds of chemically nonequivalent carbons: 2 *meta*-carbons, 2 *ortho*-carbons, 1 *para*-carbon, 1 *ipso*-carbon, and 1 carbonyl carbon.
 - Table B shows that the aromatic carbons come into resonance in the range 100-170 δ , and the carbonyl carbon comes into resonance in the range 182-215 δ .
- E. Observed spectral data for benzaldehyde:
- ^1H NMR (CDCl_3):
- ^{13}C NMR (CDCl_3):

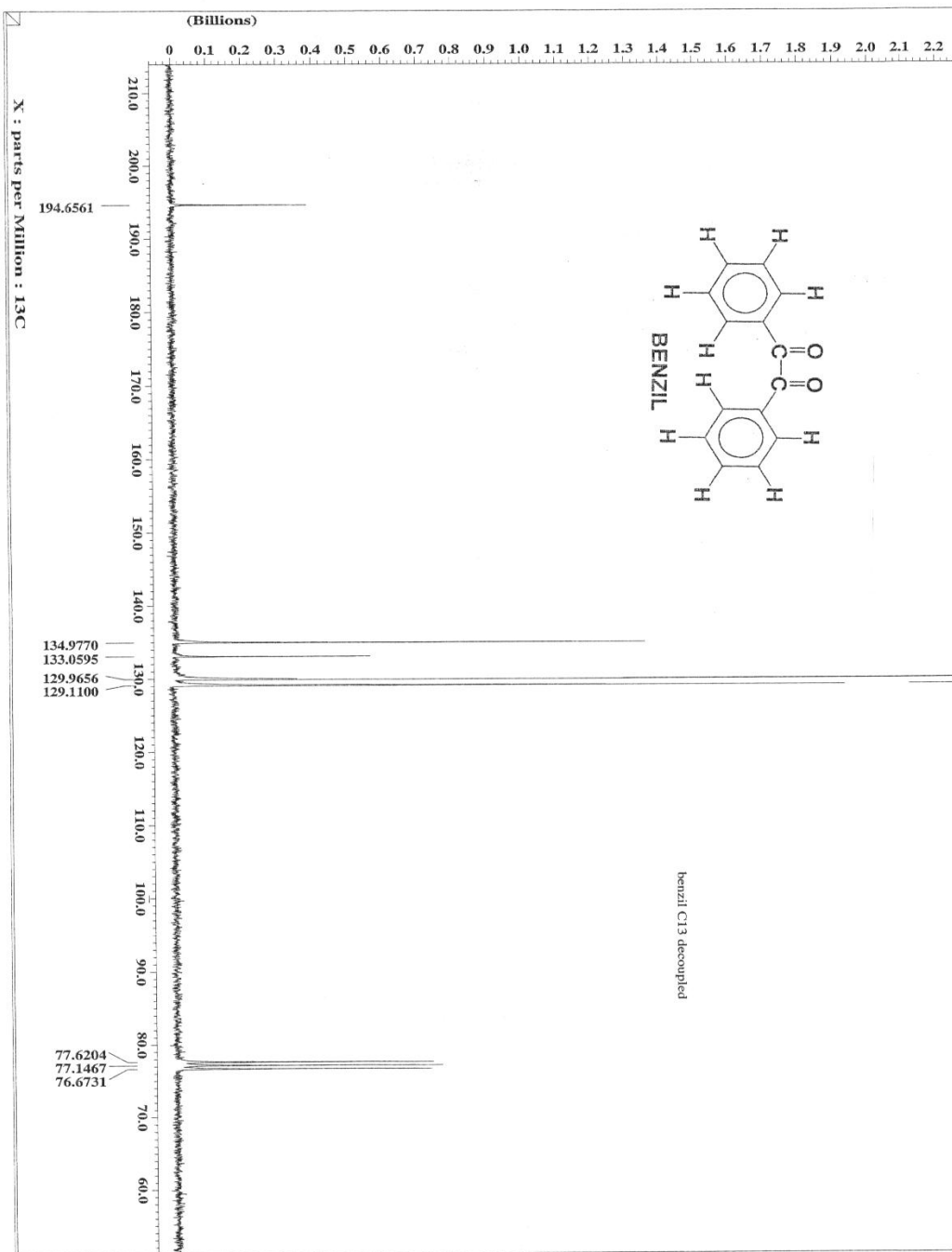


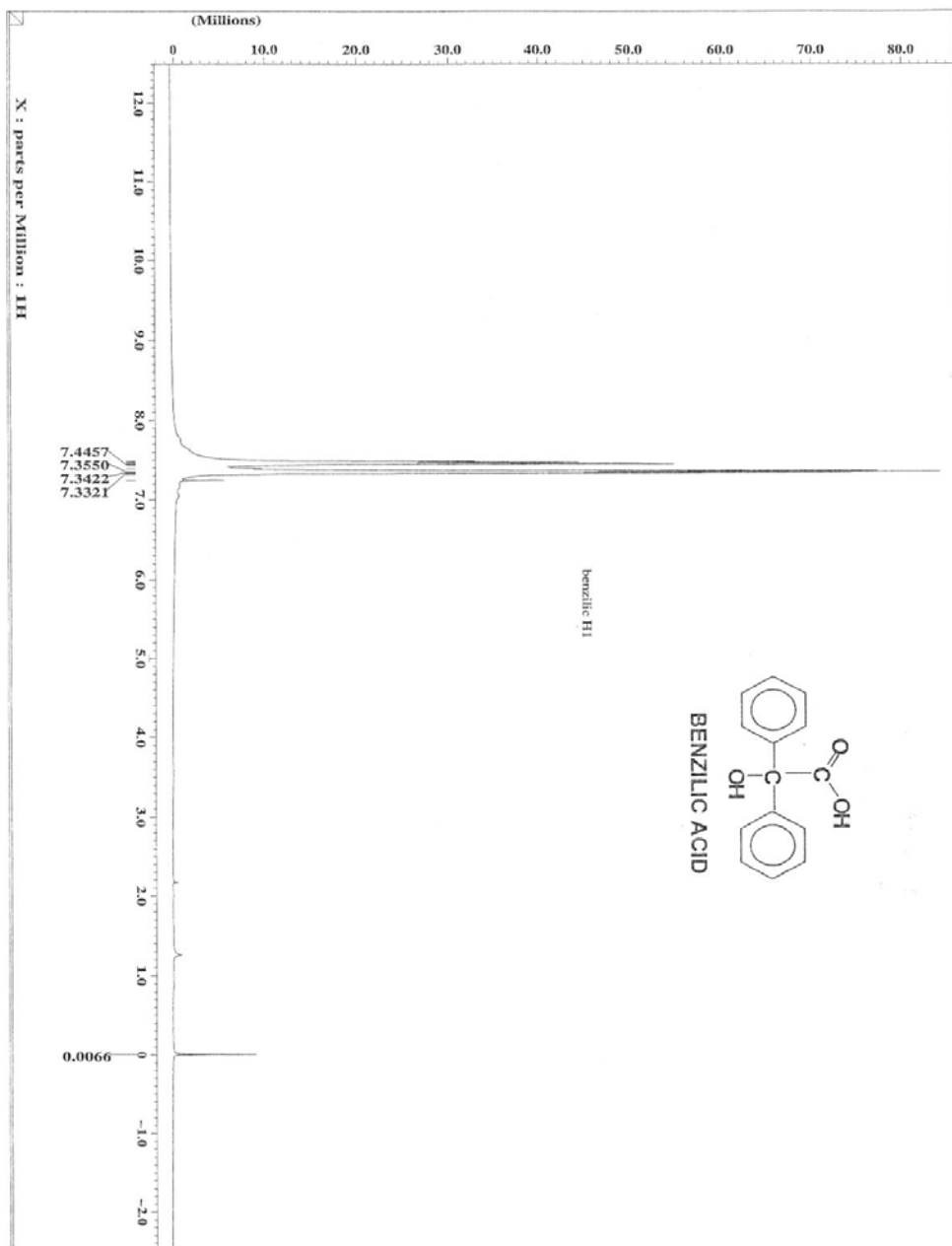


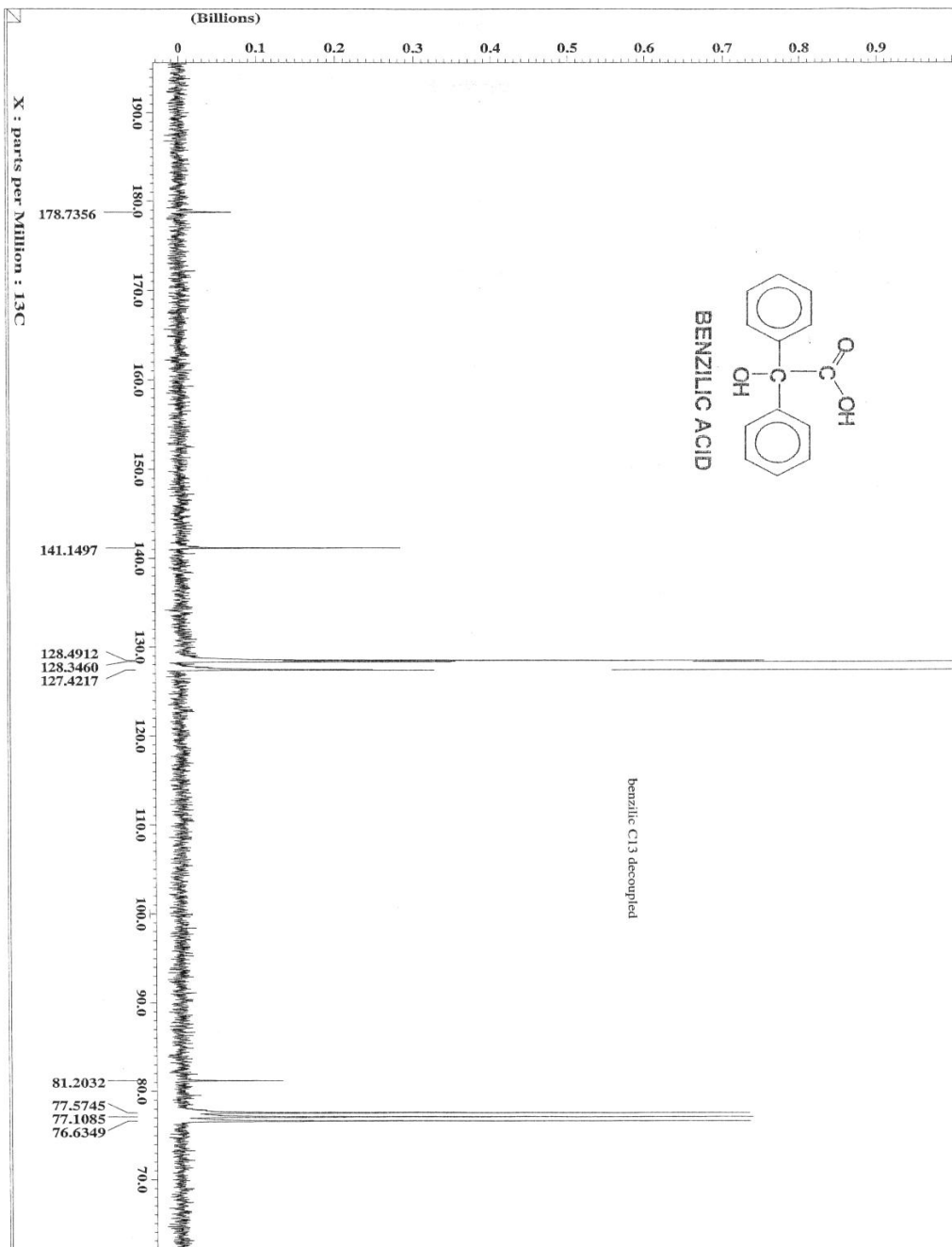












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