

HOMEWORK # 8

CHEM 121, section 1

Background and Chps. 18. Carbohydrates

Homework

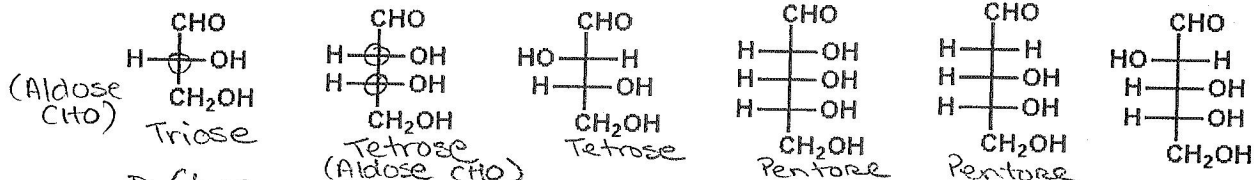
Printed Name: Kevin Key

Group Name: P

1) (2 pts) Give names of the following carbohydrates. Identify the chiral carbon atoms

Handedness = chirality. Achiral = non-handedness.

Chiral-carbon atom attached to four other (structures) groups.



a) D-Glucose b) D-erythrose c) D-threose d) D-ribose e) D-deoxy-ribose f) D-arabinose

2) (4 pts.) Circle the correct classifications that apply to the following carbohydrates.

Carbohydrate	Formula	Type	Carbons	Named	Configuration
a) $\begin{array}{c} \text{CHO} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$	$\text{C}_n\text{H}_{2n}\text{O}_n = \text{C}_3\text{H}_6\text{O}_3$	<u>Aldose</u> /ketose	<u>Triose</u> / Tetrose/ Pentose/ Hexose	Name: D-Glyceraldehyde	<u>D</u> /L
b) $\begin{array}{c} \text{CHO} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$	$\text{C}_n\text{H}_{2n}\text{O}_n = \text{C}_4\text{H}_8\text{O}_4$	<u>Aldose</u> /ketose	Triose/ <u>Tetrose</u> / Pentose/ Hexose	Name: D-threose	<u>D</u> /L
c) $\begin{array}{c} \text{CHO} \\ \\ \text{H}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ 4 OH here would have made it ribose.	$\text{C}_n\text{H}_{2n}\text{O}_n = \text{C}_5\text{H}_{10}\text{O}_4$	<u>Aldose</u> /ketose	Triose/ Tetrose/ <u>Pentose</u> / Hexose	Name: D-deoxy-ribose	<u>D</u> /L
d) $\begin{array}{c} \text{CHO} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$	$\text{C}_n\text{H}_{2n}\text{O}_n = \text{C}_6\text{H}_{12}\text{O}_6$	<u>Aldose</u> /ketose	Triose/ Tetrose/ Pentose/ <u>Hexose</u>	Name: D-glucose	<u>D</u> /L
e) $\begin{array}{c} \text{CH}_2\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ converted to ketone group.	$\text{C}_n\text{H}_{2n}\text{O}_n = \text{C}_6\text{H}_{12}\text{O}_6$	Aldose/ <u>ketose</u>	Triose/ Tetrose/ Pentose/ <u>Hexose</u>	Name: D-fructose	<u>D</u> /L

a molecule by polarizing light thru it.
* deoxy - w/o oxygen (#1e.)

e
trioses:
trioses:
erythrose

* Carbohydrates are water-soluble b/c they have an OH group. The more

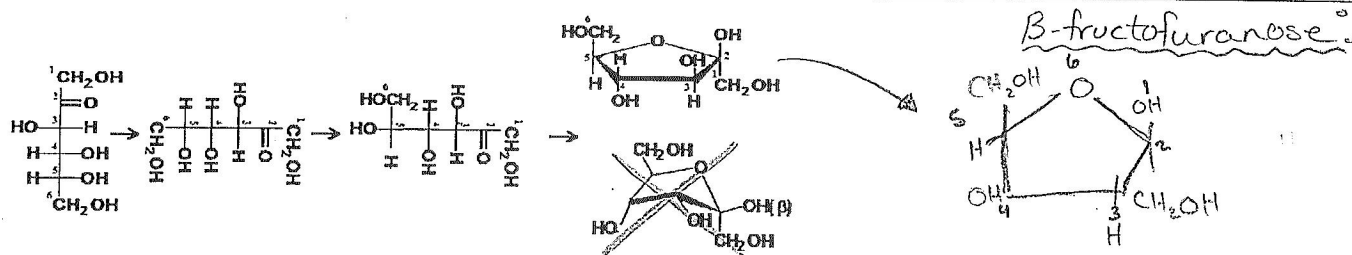
D-glucose
Right
L-glucose
Left

- 3) (3 pts.) Complete the names, circle carbon atom numbers of the OH group which is on the left and the structure for following aldohexoses.

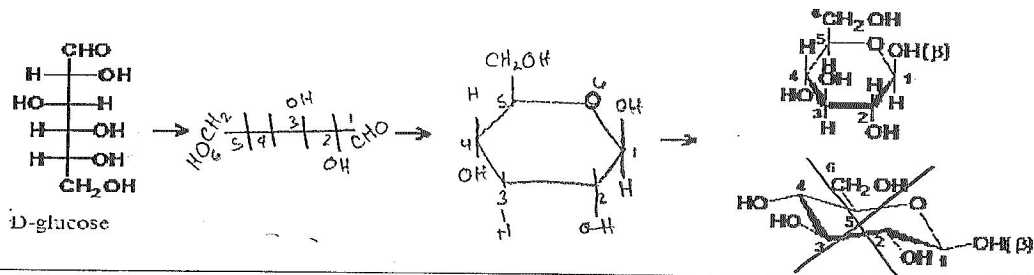
All	Altruists	Gladly	Make
Name: <u>D-Allose</u>	Name: <u>D-Allose</u>	Name: <u>D-Glucose</u>	Name: <u>D-Mannose</u>
C2/C3/C4	<u>(C2)</u> /C3/C4	C2/ <u>(C3)</u> /C4	<u>(C2)</u> / <u>(C3)</u> /C4
$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{H} - 2 - \text{OH} \\ \\ \text{H} - 3 - \text{OH} \\ \\ \text{H} - 4 - \text{OH} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $	$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{HO} - 2 - \text{H} \\ \\ \text{H} - 3 - \text{OH} \\ \\ \text{H} - 4 - \text{OH} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $	$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{H} - 2 - \text{OH} \\ \\ \text{OH} - 3 - \text{H} \\ \\ \text{H} - 4 - \text{OH} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $	$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{HO} - 2 - \text{H} \\ \\ \text{HO} - 3 - \text{H} \\ \\ \text{H} - 4 - \text{OH} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $
Gum	in	Gallon	Tanks
Name: <u>D-Gulose</u>	Name: <u>D-Idose</u>	Name: <u>D-Galactose</u>	Name: <u>D-Tallose</u>
C2/C3/ <u>(C4)</u>	<u>(C2)</u> /C3/ <u>(C4)</u>	C2/ <u>(C3)</u> / <u>(C4)</u>	<u>(C2)</u> / <u>(C3)</u> / <u>(C4)</u>
$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{H} - 2 - \text{OH} \\ \\ \text{H} - 3 - \text{OH} \\ \\ \text{OH} - 4 - \text{H} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $	$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{HO} - 2 - \text{H} \\ \\ \text{H} - 3 - \text{OH} \\ \\ \text{HO} - 4 - \text{H} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $	$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{H} - 2 - \text{OH} \\ \\ \text{HO} - 3 - \text{H} \\ \\ \text{HO} - 4 - \text{H} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $	$ \begin{array}{c} 1 \text{ CHO} \\ \\ \text{HO} - 2 - \text{H} \\ \\ \text{HO} - 3 - \text{H} \\ \\ \text{HO} - 4 - \text{H} \\ \\ \text{H} - 5 - \text{OH} \\ \\ 6 \text{ CH}_2\text{OH} \end{array} $

- 4) (3 pts.) Complete the hemiacetal structures for following carbohydrates. Circle the anomeric carbon atom.

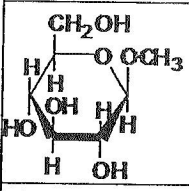
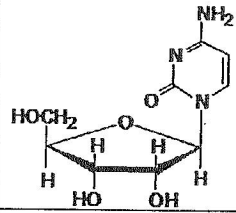
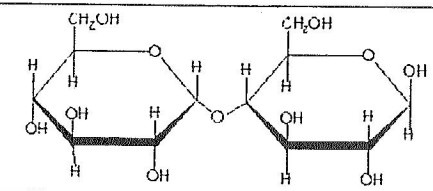
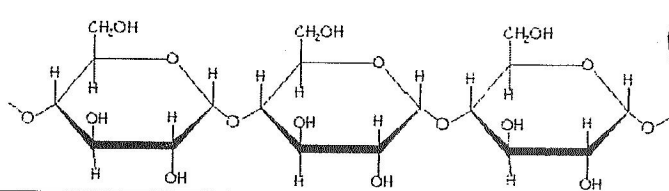
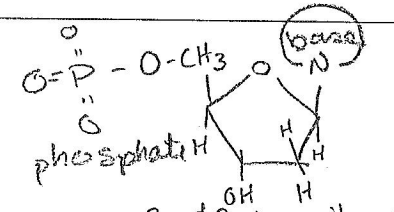
The following diagram shows how to get the cyclic hemiacetal form of the ketohexose sugar D-fructofuranose with β -anomeric



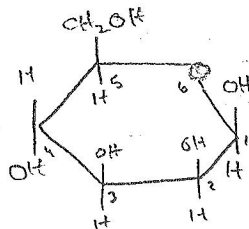
- a) Complete following diagram to get the cyclic hemiacetal form of the aldohexose sugar D-glucopyranose with β -anomeric configuration.



7) (3 pts.) Answer following question about glycosidic bond of the di-, oligo- and poly- saccharides.

a) Name of the following glucoside  <i>Methoglycoside</i>	Answer
b) Name of the following N-Glycoside  <i>N-cytosine glycoside ribose</i>	Answer
c) a) Name of the following glucoside.  <i>Maltose (alpha-1,4 glycoside)</i>	Answer
d) The following is a part of a cellulose polymeric chain. What is the type of glucosidic linkage?  <i>Amylose (alpha-1,4 linkage)</i>	Answer
e) What is DNA? Describe the components. <i>Nucleotide made up of 4 different bases. Double stranded.</i>	 <i>phosphate</i> <i>2-deoxyribose</i>
f) What is RNA? Describe the components. <i>Single stranded.</i>	

b) Draw the structure of β -D-mannopyranose



c) What is the difference between pyranose and furanose types of sugars?

Pyranose sugars have 6 carbons & furanose have 5 carbons. Hexoaldoses make pyranose. Hexoketoses make furanoses. Pentoses are all furanoses.

d) What is the difference between α and β forms of cyclic hemiacetal forms of sugars?

α is the carbon atom & anomeric is opposite to the OH group.

5) (3 pts.) Convert the hemiacetal cyclic structures (Haworth Projections) for following carbohydrates to Fischer Projections.

The following diagram shows how to get the cyclic hemiacetal form of the hexoketose sugar D-fructofuranose with β -anomeric

a) Haworth Projections of cyclic hemiacetal.	Name of the compound: (2,3) α-D-lyxofuranose	Fischer Projection of linear form:	Name of the compound: D-lyxose
b) Haworth Projections of cyclic hemiacetal. 5-carbons (Pentose)	Name of the compound: D-ribose	Fischer Projection of linear form:	Name of the compound: D-Ribose

6) (2 pts) What are the following? What is/are glucosidic linkage found in them? α or β depending on if the anomeric carbon is up or down. (1,4 α or 1,4 β are most common.)

- Amylose* *polymer of glucose*
- Starch:** (Energy storage of plants.)
 α -1,4-glycosidic links (chains)
 - Amylopectin:**
 α -1,6-glycosidic link
 - Glycogen:** made up of Amylopectin
 α -1,6 linkage at cross link chains
(energy storage of animals.)
 - Cellulose**
 β -1,4 glycosidic linkages