

HOMWORK #5

CHEM 121, section 1

Printed Name: Key

Background and Chp. 15. Introduction to aldehyde and Ketones

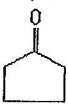
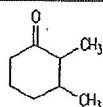
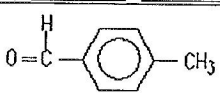
Homework Due

Group Name: 21

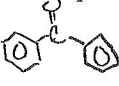
- 1) (2 pts) Assign the type of organic compound with following general condensed formula.

$\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$ a) <u>carboxylic acid</u>	$\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$ b) <u>aldehyde</u>	$\text{R}'-\overset{\text{O}}{\parallel}{\text{C}}-\text{R}$ c) <u>Ketone</u>	$\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$ d) <u>aldehyde</u>
$\begin{array}{c} \text{OH} \\ \\ \text{R}-\text{C}-\text{O}-\text{R}' \\ \\ \text{H} \end{array}$ e) <u>hemiacetal</u>	$\begin{array}{c} \text{OH} \\ \\ \text{R}-\text{C}-\text{O}-\text{R}'' \\ \\ \text{R}' \end{array}$ f) <u>hemiketal</u>	$\begin{array}{c} \text{OR}'' \\ \\ \text{R}-\text{C}-\text{O}-\text{R}' \\ \\ \text{H} \end{array}$ g) <u>acetal</u>	$\begin{array}{c} \text{OR}''' \\ \\ \text{R}-\text{C}-\text{O}-\text{R}'' \\ \\ \text{R}' \end{array}$ h) <u>hemiacetal</u>

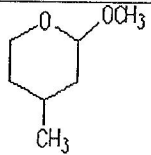
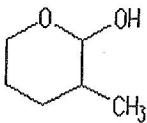
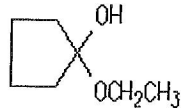
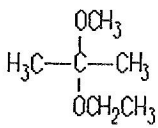
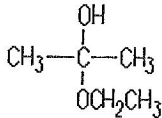
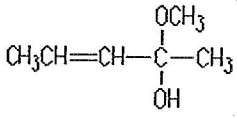
- 2) (2 pts) Names (common/IUPAC) of following aldehydes and ketones

$\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$ a) <u>acetaldehyde</u> <u>ethanal</u>	$\text{CH}_3-\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$ b) <u>propanal</u>	$\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_2-\overset{\text{CH}_3}{\text{CH}}-\text{CH}_3$ c) <u>4-methyl-2-pentanone</u>	$\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_3$ d) <u>acetone</u> <u>propanone</u>
$\text{CH}_3-\overset{\text{Cl}}{\text{CH}}-\overset{\text{Cl}}{\text{CH}}-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$ e) <u>3,6-dichlorobutanal</u>	 f) <u>cyclopentanone</u>	 g) <u>2,3-dimethylcyclohexanone</u>	 h) <u>4-methylbenzaldehyde</u>

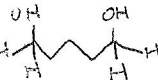
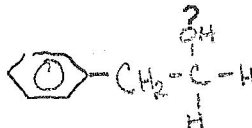
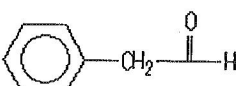
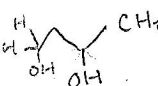
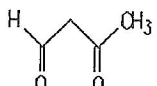
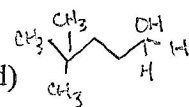
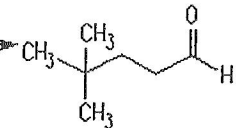
- 3) (2 pts) Draw the condensed formula of following aldehydes and ketones

a) acetaldehyde/ethanal CH_3CHO	b) ethyl methyl ketone $\text{CH}_3\text{CH}_2\overset{\text{O}}{\parallel}{\text{C}}\text{CH}_3$	c) 3-methyl-2-pentanone $\text{CH}_3\overset{\text{O}}{\parallel}{\text{C}}\text{CH}_2\text{CH}_2\text{CH}_3$	d) isopropyl methyl ketone $(\text{CH}_3)_2\text{CH}\overset{\text{O}}{\parallel}{\text{C}}\text{CH}_3$
e) 2-methylbutanal $\text{CH}_3\text{CH}_2\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{CH}}}\text{CHO}$	f) 5-methyl-3-hexanone $\text{CH}_3\text{CH}_2\overset{\text{O}}{\parallel}{\text{C}}\text{CH}_2\overset{\text{CH}_3}{\text{CH}}\text{CH}_2\text{CH}_3$	g) benzophenone 	h) acetophenone 

- 4) (2 pts) Identify each of the following compounds as a hemiacetal, hemiketal, acetal, or ketal:

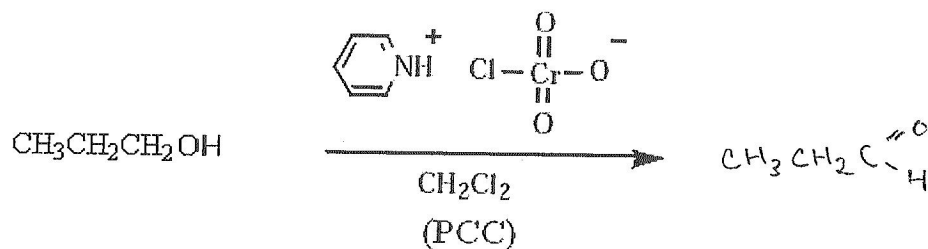
 a) <u>acetal</u>	 b) <u>hemiacetal</u>	 c) <u>hemiketal</u>
 d) <u>ketal</u>	 e) <u>hemiketal</u>	 f) <u>hemiketal</u>

- 5) (2 pts) Which alcohol would you oxidize to produce each of the following compounds?

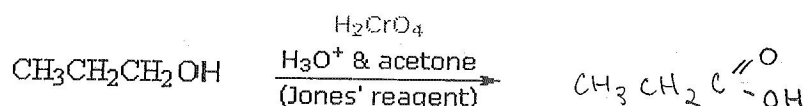
a)  ?	$\xrightarrow{\text{PCC}}$ 
b)  ?	$\xrightarrow{\text{PCC}}$ 
c)  ?	$\xrightarrow{\text{PCC}}$ 
d)  ?	$\longrightarrow$ 

6) (3 pts) Preparation of aldehydes and ketones

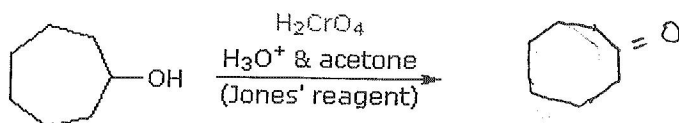
a) Control oxidation of 1^{ry} alcohol



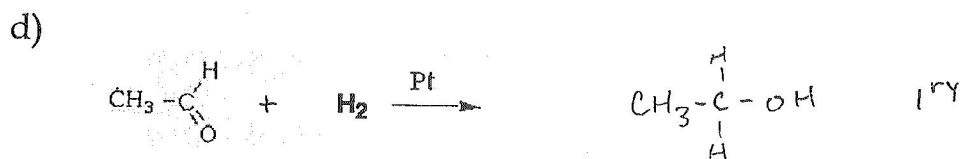
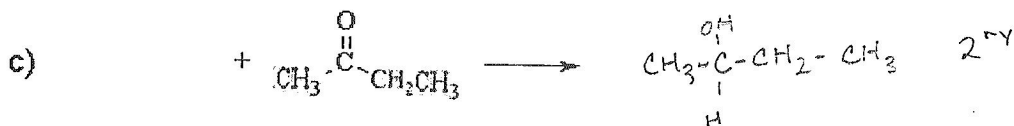
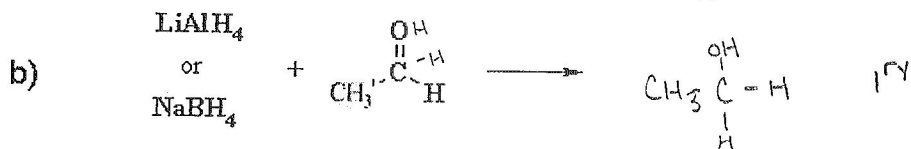
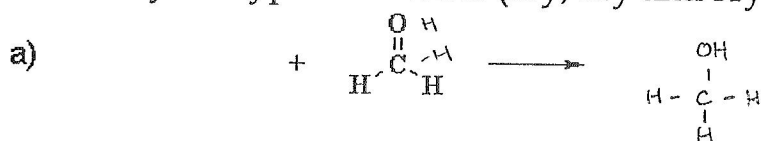
b) Complete oxidation of 1^{ry} alcohol



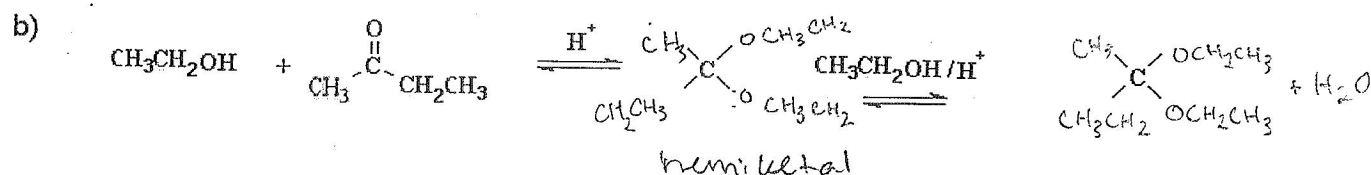
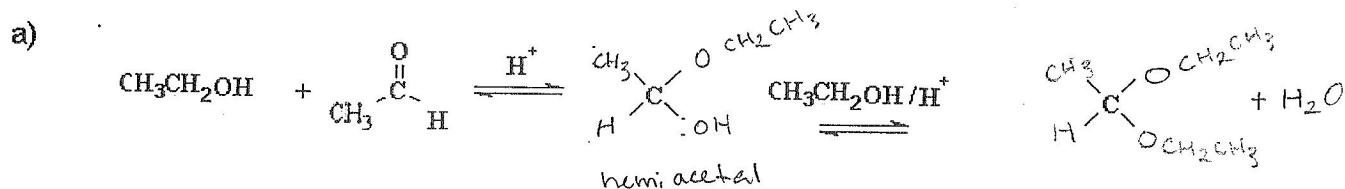
c) Oxidation of 2^{ry} alcohol



7) (2 pts) Complete the following reductions of aldehyde and ketones and identify the type of alcohols (1^{ry}, 2^{ry} and 3^{ry}) produced.

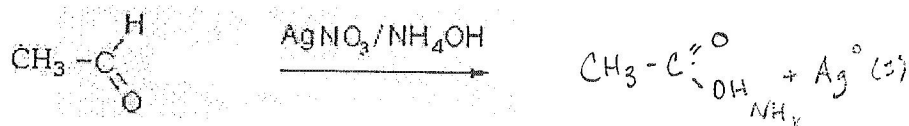


8) (2 pts) Complete the following formation reactions hemiacetal, hemiketal, acetal, or ketal:

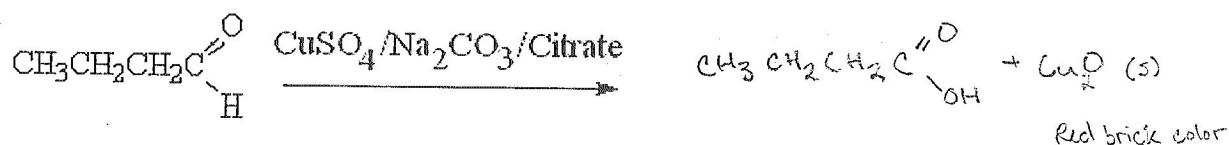


9) (3 pts) Complete reactions of following aldehydes and ketones

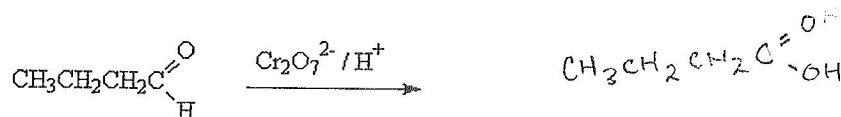
a) Tollen's Reagent: Silver mirror test:



b) Benedict's test:



c) Oxidation of aldehyde



d) Cyclic hemiacetal formation

