

HOMEWORK #10

CHEM 121, section 1

Background for the Chapter. 20. Proteins

Homework 1

Printed Name:

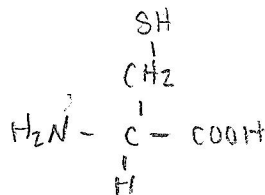
Key

Group Name:

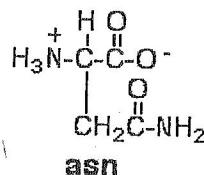
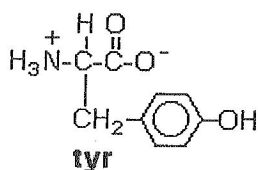
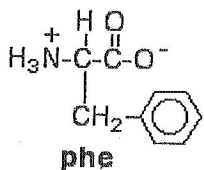
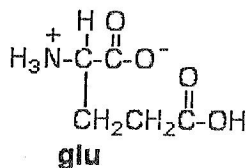
- 1) (3 pts) Give name, abbreviation and types (neutral, polar, non-polar, basic and acidic).

non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_3 \end{array}$ alanine	polar basic $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{CH}_2\text{CH}_2\text{NHC(=NH)NH}_2 \end{array}$ Arginine	polar neutral $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{C(=O)NH}_2 \end{array}$ asparagine	polar acidic $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{C(=O)OH} \end{array}$ aspartic acid
polar neutral $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{SH} \end{array}$ cysteine	glycine non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{H} \end{array}$ glycine	polar neutral $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{CH}_2\text{C(=O)NH}_2 \end{array}$ Glutamine	polar acidic $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{CH}_2\text{C(=O)OH} \end{array}$ Glutamic acid
polar basic $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2 \\ \\ \text{HN} \end{array}$ Histidine	non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH} \\ / \quad \backslash \\ \text{CH}_3 \quad \text{CH}_2\text{CH}_3 \end{array}$ Isoleucine	non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_3 \end{array}$ Leucine	polar basic $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2 \end{array}$ Lysine
non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{CH}_2\text{SCH}_3 \end{array}$ Methionine	non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2 - \text{C}_6\text{H}_5 \end{array}$ Phenylalanine	non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH} - \text{C} - \text{H} \\ \\ \text{C}_5\text{H}_9 \end{array}$ Proline	polar neutral $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2\text{OH} \end{array}$ Serine
Threonine $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}(\text{OH})\text{CH}_3 \end{array}$ polar neutral	Tryptophan $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2 \\ \\ \text{C}_8\text{H}_6\text{N} \end{array}$ non polar	Polar neutral $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}_2 - \text{C}_6\text{H}_4 - \text{OH} \end{array}$ Tyrosine	non polar $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{NH}_2 - \text{C} - \text{H} \\ \\ \text{CH}(\text{CH}_3)_2 \end{array}$ Valine

2) Draw the optical and L isomers for: cys.



3) Use the following amino acids to answer the questions below:



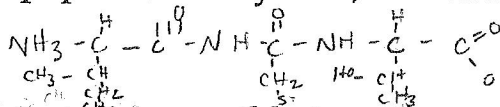
- Which amino acid is most polar? *tyr*
- Which amino acid is most non-polar? *phe*
- Which amino acid gives an acidic solution? *glu*
- Which amino acid gives a basic solution? *asn*

4) Draw the following:

a) Dipeptide bond between **ala** and **asp**, and identify C- and N-terminal.

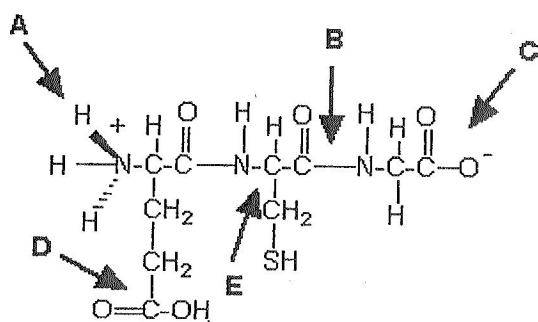


b) Tripeptide, **ile-cys-thr**, and identify N- (left) and C-terminal(right).



c) How many possible isomers are in the tripeptide formed with **ile**, **cys** and **thr**?
Come up with a formula for linear chain with "n" amino acids.

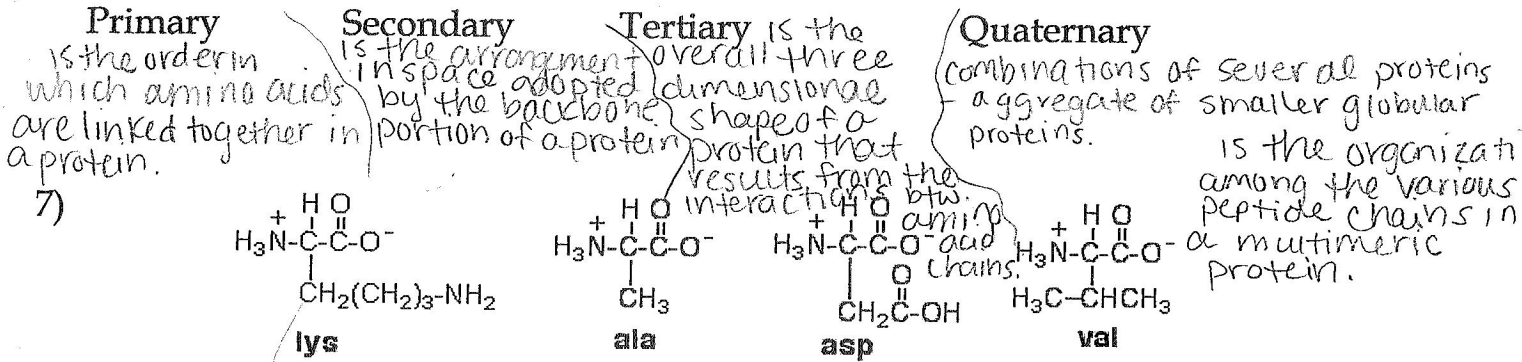
d) Give the IUPAC name of the tripeptide with the sequence, **ile-cys-thr**.



5) Use the structure to answer the questions below

- Which letter arrow points the end of the peptide that is the "amine" end-N-terminal? *A*
- Which letter arrow points the end of the peptide that is the "carboxyl" end, C-terminal? *D*
- Which letter arrow points to an amide or peptide bond? *C*

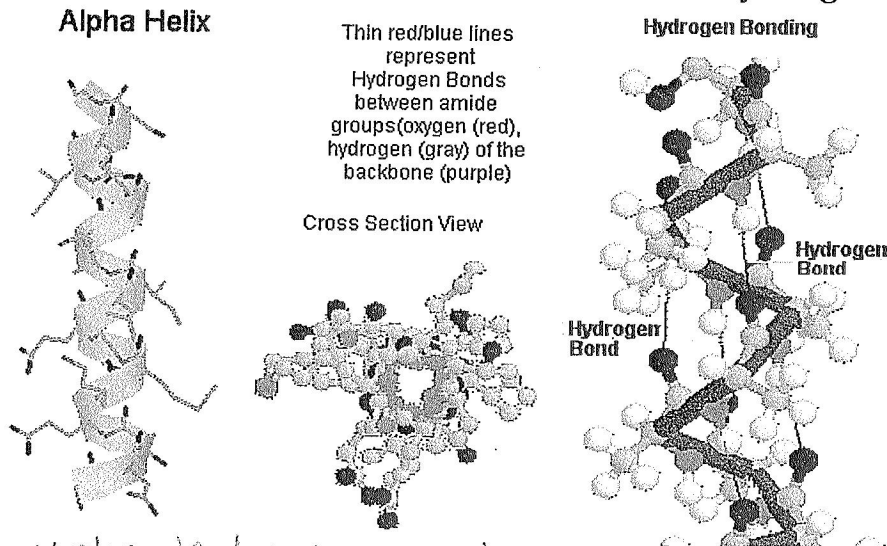
- 6) Explain the differences between primary, secondary, tertiary, and quaternary protein structures by giving brief definitions of each. What types of bonding are used in each?



Use the above structures to answer the questions below:

- a. Which two amino acids may link in a salt bridge in tertiary protein structure?
lys, ala, asp
- b. Which two amino acids may link in hydrophobic interactions in tertiary protein structure?
ala, val
- c. Which two amino acids may link in hydrogen bonding interactions in tertiary protein structure?
lys and asp

- 8) Explain the difference between the alpha helix and the beta pleated sheet protein structures. What are the differences in the hydrogen bonding?



an alpha helix structure is a protein secondary structure in which a single protein chain adopts a shape that resembles a coiled spring, with the coil configuration maintained by hydrogen bonds.

A beta pleated sheet structure is a protein secondary structure in which two fully extended protein chain segments in the same or different molecules are held together by hydrogen bonds.