

Chapter 6

Calculations: Formula Masses, Moles, and Chemical Equations

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Chapter 6-1

Diamond is a covalent compound with many carbon atoms

→ **CO 6.1**
Close-up of cut diamonds



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Chapter 6-2

Molar mass: The mass units



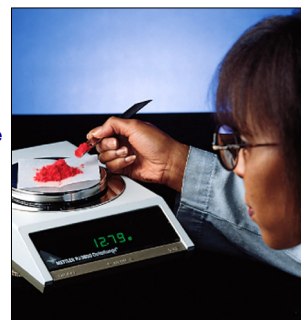
← Fig. 6.1
Oranges may be bought in units of mass or units of amount.

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Chapter 6-3

First chemical compound need to be weighed

→ Fig. 6.2
A basic process in chemical laboratory work is determining the mass of a substance.



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Chapter 6-4

Mole is a counting unit used by chemists



← Fig. 6.3
Everyday counting units.

Unlike a dozen Avogadro's number is large: 6.0022×10^{23}



→ Fig. 6.4
Amadeo Avogadro was the first scientist to distinguish between atoms and molecules.

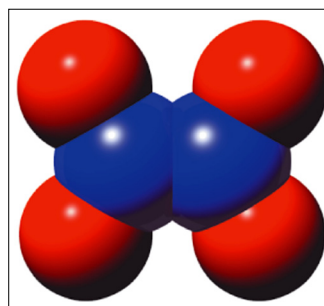
Edgar Fahs Smith Collection, University of Pennsylvania

Formula Masses in grams gives a Mole of a chemical compound



Fig. 6.5
The mass of a mole depends on the substance.

To get the Formula Masse first find the types of atoms in the compound



← Fig. 6.6
A computer-generated model of the molecular compound N_2O_4 .

Moles multiplied by Avogadro's number gives number of atoms and molecules

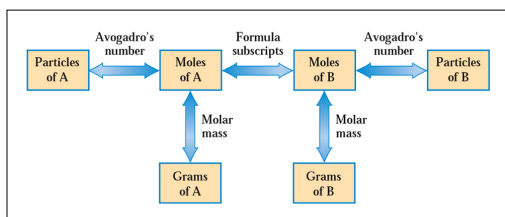


Fig. 6.7

In solving chemical-formula-based problems, the only "transitions" allowed are those between quantities (boxes) connected by arrows.

Law of conservation of mass/atoms

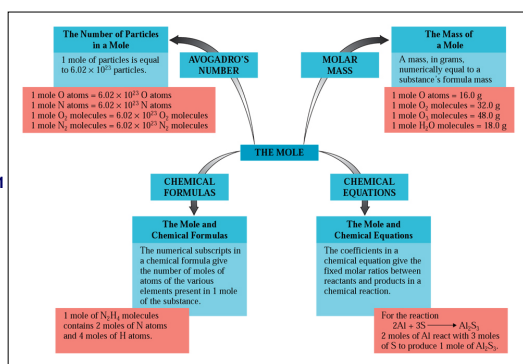
Fig. 6.8

When 16.90g of the compound CaS is decomposed into its constituent elements, the Ca and S produced has an identical mass of 16.90g.



Conversion factors in chemical calculations: Stoichiometry

→ CAG 6.1



Mole conversion to grams and particles

- Moles multiplied by Avogadro's number gives number of atoms and molecules ($\text{mole} \times N = \text{particles}$)
- Mole multiplied formula mass gives grams. ($\text{mole} \times \text{F.M.} = \text{grams}$)
- Grams divided by formula mass gives moles. ($\text{grams} \div \text{F.M.} = \text{mole}$)

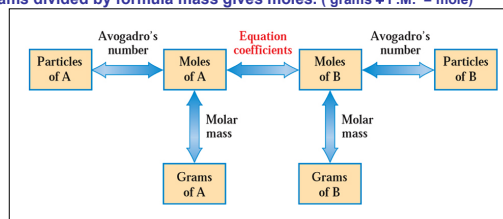


Fig. 6.9 In solving chemical-equation-based problems, the only "transitions" allowed are those between quantities (boxes) connected by arrows.

What does 16-20-0 means in fertilizer?

NITROGEN 16.0% PHOSPHATE 20.0%

→ CC 6.1



Air bag should have right amount reactants to produce gas or baby will crush.



← Fig. 6.10
Testing apparatus for
measuring the effects
of air bag deployment.

Courtesy of Daimler Chrysler Corporation